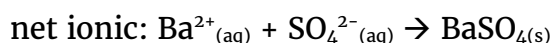
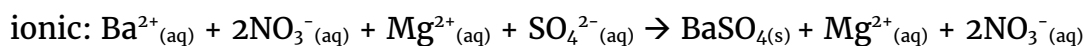
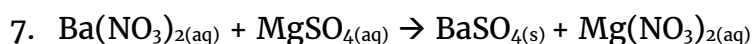
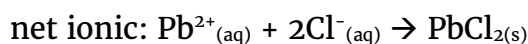
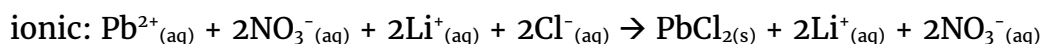
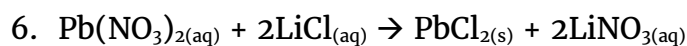
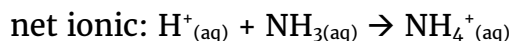
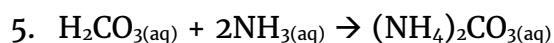
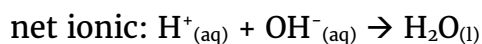
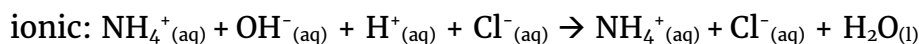
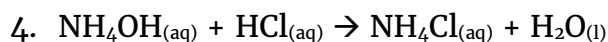
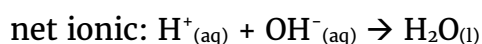
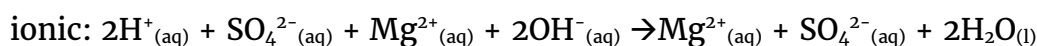
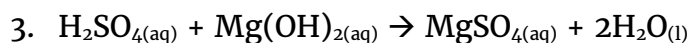
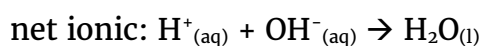
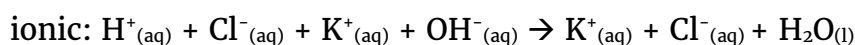
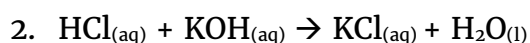
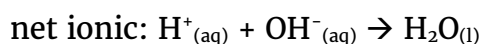
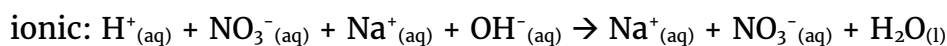
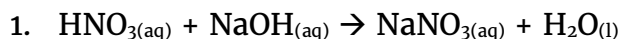


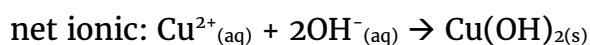
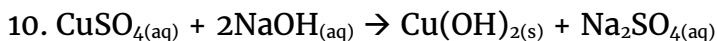
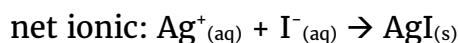
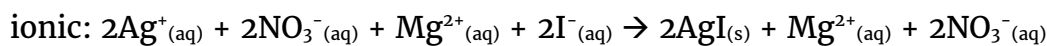
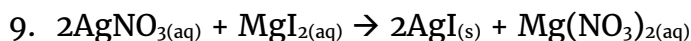
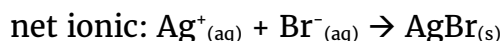
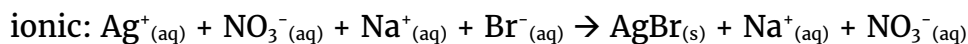
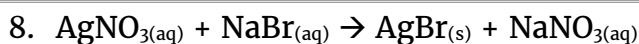
Ionic equations

NB: 'net ionic' means the ionic equation with spectator ions cancelled.

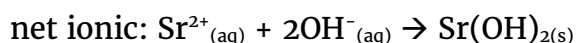
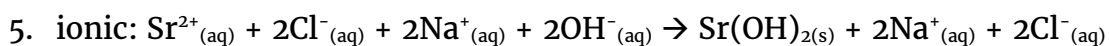
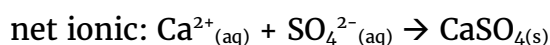
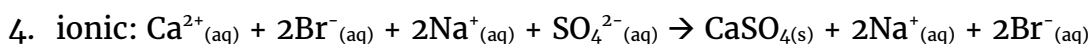
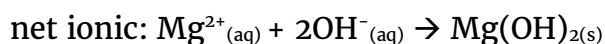
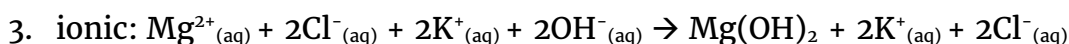
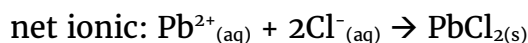
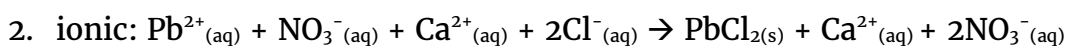
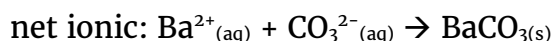
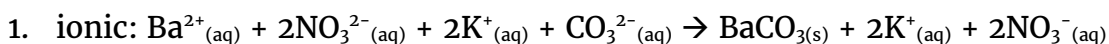
Equations with state symbols included



9-1 GCSE Chemistry Ionic Equations – Answers



Using the solubility rules



Writing ionic equations from descriptions

